

8 Most Missed Questions from the Final Exam*

#1 For the molecule CH ₃ , which atom goes in the center of the molecule and why?	#2 Name for SrCl ₂ is:	#4 Which of the following are incorrect? A) O, halogen B) K, alkali metal C) Ne, noble gas D) Ba, alkaline earth metal	#5 The correct formula for nickel(III) phosphate is what?
	#3 Name for FeO is:		
#6 Which of the following atoms cannot break the octet rule in a molecule? A) B B) P C) N D) S	#7 What is the electron configuration for Mn?		
	#8 Convert: 6.59 m/s = ___ mi/hr (1mi=5280ft, 1ft=12in, 39.37in=1m, 60min=1hr)		

*In no particular order

8 Most Missed Questions from the Final Exam*

#1 For the molecule CH ₃ , which atom goes in the center of the molecule and why?	#2 Name for SrCl ₂ is:	#4 Which of the following are incorrect? A) O, halogen B) K, alkali metal C) Ne, noble gas D) Ba, alkaline earth metal	#5 The correct formula for nickel(III) phosphate is what?
	#3 Name for FeO is:		
#6 Which of the following atoms cannot break the octet rule in a molecule? A) B B) P C) N D) S	#7 What is the electron configuration for Mn?		
	#8 Convert: 6.59 m/s = ___ mi/hr (1mi=5280ft, 1ft=12in, 39.37in=1m, 60min=1hr)		

*In no particular order